***Introduction to General, Organic & Biological Chemistry***

**Chapter 2 Chemistry and Measurement**

2.1 Multiple-Choice Questions

1) The amount of space occupied by a substance is its \_\_\_\_\_\_\_\_.

A) mass

B) density

C) weight

D) length

E) volume

Answer: E

2) Which of the following is the basic unit of volume in the metric system?

A) liter

B) kilogram

C) meter

D) centimeter

E) gram

Answer: A

3) Which of the following is a measurement of mass in the metric system?

A) milliliter

B) centimeter

C) kilogram

D) Celsius

E) meter

Answer: C

4) A value of 25 °C is a measurement of \_\_\_\_\_\_\_\_.

A) length

B) volume

C) temperature

D) mass

E) density

Answer: C

5) A value of 36 mL is a measure of \_\_\_\_\_\_\_\_.

A) density

B) mass

C) temperature

D) volume

E) length

Answer: D

6) A value of 345 mm is a measure of \_\_\_\_\_\_\_\_.

A) density

B) mass

C) temperature

D) volume

E) length

Answer: E

7) The measurement of the gravitational pull on an object is its \_\_\_\_\_\_\_\_.

A) volume

B) weight

C) mass

D) length

E) size

Answer: B

8) Which of the following measurements has three significant figures?

A) 0.005 m

B) 510 m

C) 0.510 m

D) 0.051 m

E) 5100 m

Answer: C

9) Which of the following numbers contains the designated correct number of significant figures?

A) 0.04300 5 significant figures

B) 0.00302 2 significant figures

C) 156 000 3 significant figures

D) 1.04 2 significant figures

E) 3.0650 4 significant figures

Answer: C

10) The number of significant figures in the measurement of 45.030 mm is \_\_\_\_\_\_\_\_.

A) none

B) three

C) four

D) five

E) six

Answer: D

11) How many significant figures are in the number 0.00208?

A) six

B) two

C) three

D) four

E) five

Answer: C

12) Which of the following examples illustrates a number that is correctly rounded to three significant figures?

A) 4.05438 g to 4.054 g

B) 0.03954 g to 0.040 g

C) 103.692 g to 103.7 g

D) 109 526 g to 109 500 g

E) 20.0332 g to 20.0 g

Answer: E

Learning Obj.: 2.3

13) A calculator answer of 423.6059 must be rounded off to three significant figures. What answer is reported?

A) 423

B) 424

C) 1.7420

D) 423.6

E) 423.7

Answer: B

Learning Obj.: 2.3

14) Which of the answers for the following conversions contains the correct number of significant figures?

A) 2.543 m ×  = 100.1942 in.

B) 2 L ×  = 2.12 qt

C) 24.95 min ×  = 0.4158 hr

D) 12.0 ft ×  ×  = 370 cm

E) 24.0 kg ×  = 11 lb

Answer: C

Learning Obj.: 2.3

15) What is the correct answer for the calculation of a volume (in mL) with measured numbers ?

A) 0.22 mL

B) 0.223 mL

C) 57 mL

D) 14 mL

E) 14.3 mL

Answer: A

Learning Obj.: 2.3

16) When 2610 + 11.7 + 0.22 are added, the answer to the correct number of decimal places is \_\_\_\_\_\_\_\_.

A) 2621.92

B) 2621.9

C) 2621

D) 2620

E) 2600

Answer: D

Learning Obj.: 2.3

17) What is the answer, with the correct number of decimal places, for this problem?

 4.392 g + 102.40 g + 2.51 g =

A) 109.302 g

B) 109 g

C) 109.3 g

D) 109.30 g

E) 110 g

Answer: D

Learning Obj.: 2.3

18) The correct answer for the addition of 7.5 g + 2.26 g + 1.311 g + 2 g is \_\_\_\_\_\_\_\_.

A) 13.071 g

B) 13 g

C) 13.0 g

D) 10 g

E) 13.1 g

Answer: B

Learning Obj.: 2.3

19) In which of the following is the metric unit paired with its correct abbreviation?

A) microgram / mg

B) milliliter / mL

C) centimeter / km

D) kilogram / cg

E) gram / gm

Answer: B

20) Which of the following is the largest unit?

A) millimeter

B) micrometer

C) meter

D) decimeter

E) kilometer

Answer: E

21) What is the metric relationship between grams and micrograms?

A) 1 g = 100 μg

B) 1 g = 1 000 000 μg

C) 1 g = 0.000 001 μg

D) 1 g = 1000 μg

E) 1 g = 0.001 μg

Answer: B

22) Which of the following is the smallest unit?

A) gram

B) milligram

C) kilogram

D) decigram

E) microgram

Answer: E

23) The cubic centimeter (c or cc) has the same volume as a \_\_\_\_\_\_\_\_.

A) cubic inch

B) cubic liter

C) milliliter

D) centimeter

E) cubic decimeter

Answer: C

24) What is the conversion factor for the relationship between millimeters and centimeters?

A) 1 mm/1 cm

B) 10 mm/1 cm

C) 1 cm/1 mm

D) 100 mm/1 cm

E) 10 cm/1 mm

Answer: B

Learning Obj.: 2.5

25) Which of the following conversion factors is a measured number?

A) 10 cm/dm

B) 12 in/ft

C) 16 oz/lb

D) 25 miles/gallon

E) 12 eggs/dozen

Answer: D

Learning Obj.: 2.5

26) A conversion factor set up correctly to convert 15 inches to centimeters is \_\_\_\_\_\_\_\_.

A) 100 cm/1 m

B) 1 inch/2.54 cm

C) 1 cm/10 mm

D) 2.54 cm/1 inch

E) 10 cm/1 inch

Answer: D

Learning Obj.: 2.5

27) 9.31 g is the same mass as \_\_\_\_\_\_\_\_.

A) 931 μg

B) 931 kg

C) 93.1 cg

D) 9310 mg

E) 0.0931 dg

Answer: D

Page Ref: 2.6

Learning Obj.: 2.6

28) According to the United States Food and Drug Administration, the recommended daily requirement of protein is 44 g. This is \_\_\_\_\_\_\_\_ of protein.

A) 1248.5 oz

B) 320 000 oz

C) 1.6 oz

D) 0.0605 oz

E) 150 000 oz

Answer: C

Page Ref: 2.6

Learning Obj.: 2.6

29) 1.00 pint of milk has a volume of how many milliliters?

A) 473 mL

B) 530. mL

C) 1000 mL

D) 1890 mL

E) 106 mL

Answer: A

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Learning Obj.: 2.6

30) Which of the following setups would convert centimeters to feet?

A) cm ×  × 

B) cm ×  × 

C) cm ×  × 

D) cm ×  × 

E) cm ×  × 

Answer: C

Page Ref: 2.6

Learning Obj.: 2.6

31) How many pounds are in 3.5 kg?

A) 7.7 lb

B) 1.59 lb

C) 0.629 lb

D) 1.6 lb

E) 7.70 lb

Answer: A

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Learning Obj.: 2.6

32) How many liters of soft drink are there in 5.25 qt?

A) 4950 L

B) 55.7 L

C) 4.95 L

D) 5.57 L

E) 5.0 L

Answer: C

Page Ref: 2.6

Learning Obj.: 2.6

33) 5.21 cm is the same length as \_\_\_\_\_\_\_\_.

A) 0.0521 m

B) 52.1 dm

C) 5.21 mm

D) 0.00521 km

E) 5210 m

Answer: A

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Learning Obj.: 2.6

34) How many centimeters are there in 57.0 in.?

A) 22 cm

B) 0.0445 cm

C) 145 cm

D) 22.4 cm

E) 140 cm

Answer: C

Page Ref: 2.6

Learning Obj.: 2.6

35) What is 6.5 m converted to inches?

A) 1700 in

B) 1651 in

C) 39 in

D) 260 in

E) 255.9 in

Answer: D

Page Ref: 2.6

Learning Obj.: 2.6

36) How many kilograms are in 30.4 lb?

A) 13.8 kg

B) 14 kg

C) 67 kg

D) 66.88 kg

E) 66.9 kg

Answer: A

Page Ref: 2.6

Learning Obj.: 2.6

37) A dose of aspirin of 5.0 mg per kilogram of body weight has been prescribed to reduce the fever of an infant weighing 8.5 pounds. The number of milligrams of aspirin that should be administered is \_\_\_\_\_\_\_\_.

A) 19 mg

B) 53 mg

C) 1.6 mg

D) 5.0 mg

E) 0.59 mg

Answer: A

Page Ref: 2.6

Learning Obj.: 2.6

38) A doctor's order is 0.125 g of ampicillin. The liquid suspension on hand contains 250 mg/5.0 mL. How many milliliters of the suspension are required?

A) 0.0025 mL

B) 3.0 mL

C) 2.5 mL

D) 6.3 mL

E) 0.0063 mL

Answer: C

Page Ref: 2.6

Learning Obj.: 2.6

39) Which of the following measurements are not equivalent?

A) 25 mg = 0.025 g

B) 183 L = 0.183 kL

C) 150 ms = 0.150 s

D) 84 cm = 8.4 mm

E) 24 dL = 2.4 L

Answer: D

Page Ref: 2.6

Learning Obj.: 2.6

40) Which one of the following substances will float in gasoline, which has a density of 0.74 g/mL? The density of each substance is shown in parentheses.

A) table salt (D = 2.16 g/mL)

B) balsa wood (D = 0.16 g/mL)

C) sugar (D = 1.59 g/mL)

D) aluminum (D = 2.70 g/mL)

E) mercury (D = 13.6 g/mL)

Answer: B

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41) What is the mass of 2.00 L of an intravenous glucose solution with a density of 1.15 g/mL?

A) 0.0230 kg

B) 2.30 kg

C) 1.15 kg

D) 0.0150 kg

E) 0.575 kg

Answer: B

Page Ref: 2.7

42) Mercury has a specific gravity of 13.6. How many milliliters of mercury have a mass of 0.35 kg?

A) 0.0257 mL

B) 0.026 mL

C) 25.7 mL

D) 26 mL

E) 4760 mL

Answer: D

Page Ref: 2.7

43) What is the density of a substance with a mass of 45.00 g and a volume of 26.4 mL?

A) 1.70 g/mL

B) 1.7 g/mL

C) 0.59 g/mL

D) 0.587 g/mL

E) 45.0 g/mL

Answer: A

Page Ref: 2.7

44) What is the mass of 53 mL of ethanol, which has a density of 0.79 g/mL?

A) 67.1 g

B) 41.9 g

C) 42 g

D) 67 g

E) 53 g

Answer: C

Page Ref: 2.7

45) A liquid has a volume of 34.6 mL and a mass of 46.0 g. What is the density of the liquid?

A) 1.00 g/mL

B) 1.33 g/mL

C) 0.752 g/mL

D) 1330 g/mL

E) 0.663 g/mL

Answer: B

Page Ref: 2.7

46) The density of a solution is 1.18 g/mL. Its specific gravity is \_\_\_\_\_\_\_\_.

A) 11.8

B) 0.118

C) 0.847

D) 1.18

E) 1.2

Answer: D

Page Ref: 2.7

47) Diamond has a density of 3.52 g/mL. What is the volume in cubic centimeters of a diamond with a mass of 15.1 g?

A) 4.3 

B) 4.29 

C) 0.233 

D) 53 

E) 53.2 

Answer: B

Page Ref: 2.7

48) The ratio of the mass of a substance to its volume is its \_\_\_\_\_\_\_\_.

A) specific gravity

B) density

C) buoyancy

D) weight

E) conversion factor

Answer: B

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49) A nugget of gold with a mass of 521 g is added to 50.0 mL of water. The water level rises to a volume of 77.0 mL. What is the density of the gold?

A) 10.4 g/mL

B) 6.77 g/mL

C) 1.00 g/mL

D) 0.0518 g/mL

E) 19.3 g/mL

Answer: E

Page Ref: 2.7

50) A 50.0 mL urine sample has a mass of 50.7 g. The specific gravity of the urine is \_\_\_\_\_\_\_\_.

A) 1.014 g/mL

B) 0.986 g/L

C) 1.01

D) 0.986

E) 50.7

Answer: C

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2.2 Short Answer Questions

*Round off each of the following to three significant figures.*

1) 504.85

Answer: 505

2) 8.3158

Answer: 8.32

3) 25 225

Answer: 25 200

4) 6.3477 × 

Answer: 6.35 × 

5) 399870

Answer: 4.00 × 

6) 58.5422

Answer: 58.5

7) 0.003 4088

Answer: 0.00341

*State the number of significant figures in each of the following measurements.*

8) 0.705 m

Answer: 3

9) 680 000 km

Answer: 2

10) 0.008090 cm

Answer: 4

11) 28.050 km

Answer: 5

12) 0.0005 L

Answer: 1

13) 75.00 m

Answer: 4

14) 2.043 ×  mm

Answer: 4

15) 6.1 ×  mL

Answer: 2

16) 9.00 ×  g

Answer: 3

17) If there are 14 books on the shelf, 14 is a(n) \_\_\_\_\_\_\_\_ number.

Answer: exact

2.3 True/False Questions

1) The basic unit of mass in the metric system is the pound.

Answer: FALSE

2) The liter is a unit of volume in the metric system.

Answer: TRUE

3) The number 0.0500 has four significant figures.

Answer: FALSE

4) The number 650 000 has two significant figures.

Answer: TRUE

5) When the measured number 0.0090 is multiplied by the measured number 87.10, the answer has two significant figures.

Answer: TRUE

Learning Obj.: 2.3

6) When the measured number 675 is added to the measured number 87.10, the answer should be rounded to the ones place.

Answer: TRUE

Learning Obj.: 2.3

7) A μg is larger than a mg.

Answer: FALSE

8) There are 1000 μg in a mg.

Answer: TRUE

9) A cubic centimeter is a unit of length.

Answer: FALSE

10) 1 kg equivalent to 1000 mg.

Answer: TRUE

Learning Obj.: 2.5

11) 1 mL is equivalent to 1000 L.

Answer: FALSE

Learning Obj.: 2.5

12) The density of water is 1 kg/mL.

Answer: FALSE

Page Ref: 2.7

13) Specific gravity has no units.

Answer: TRUE

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2.4 Matching Questions

*Match the type of measurement to the unit given below.*

A) mass

B) volume

C) temperature

D) length

E) density

1) milliliter

2) mm

3) gram

4) 125 K

5) kilometer

6) milligram

Answers: 1) B 2) D 3) A 4) C 5) D 6) A

*Are the numbers in each of the following statements measured or exact?*

A) exact

B) measured

7) In the U.S. system there are 12 inches in one foot.

8) The patient's blood sugar level is 350 mg/dL.

9) There are 452 pages in a book.

10) The rabbit weighs 2.5 pounds.

11) 1L is equal to 1.06 quarts.

12) There are 100 capsules in the bottle.

13) The patient's temperature is 100.1 °F.

14) I lost 14 pounds on my diet last month.

Answers: 7) A 8) B 9) A 10) B 11) B 12) A 13) B 14) B

*Select the correct prefix to complete the equality.*

A) 10

B) 1000

C) 1

D) 100

15) 1 mL = \_\_\_\_\_\_\_\_ L

16) 1 m = \_\_\_\_\_\_\_\_ mm E) 0.001

17) 1 cm = \_\_\_\_\_\_\_\_ mm

18) 1 dL = \_\_\_\_\_\_\_\_ mL

19) 1 mL = \_\_\_\_\_\_\_\_ cm3

20) 1 kg = \_\_\_\_\_\_\_\_ g

Answers: 15) E 16) B 17) A 18) D 19) C 20) B